

Reactivity of 1,2-diaza-1,3-dienes with azomethine ylides: [3+4] versus [3+2] cycloadditions

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Dedication ((optional))

Abstract: The multicomponent 1,3-dipolar cycloaddition of different 1,2-diaza-1,3-dienes with in situ-generated azomethine ylides produces 1,2,4-triazepines or pyrrolidines by means of [3+4] or [3+2] cycloadditions, respectively. The regioselectivity is controlled by the electron-withdrawing group bound to the azo-moiety of the 1,2-

diaza-1,3-diene which promotes exclusively the [3+4] cycloaddition. When the electron-withdrawing group is replaced with a phenyl group only a [3+2] cycloaddition occurs.

Introduction

1,3-Dipolar cycloaddition^[1] (1,3-DC) is a fascinating stereocontrolled reaction where a 1,3-dipole and an alkene, allene, or alkyne react to give a five-membered heterocyclic ring allowing the presence of large series of functional groups.^[2] These multicomponent,^[3] atom economy,^[4] catalyzed^[5] 1,3-DCs constitute a clear trend of innovation in both laboratory and industrial scale, cycloadditions being the most frequently employed.^[6] Some of us used this methodology in several multicomponent 1,3-DCs of diethyl aminomalonate or α -amino esters with 2-oxoaldehydes and dipolarophiles.^[7, 8, 9] At this moment, the cycloaddition between azomethine ylides and electrophilic alkenes is the most frequently reported version due to its high regio- and diastereoselection.^[10] The azomethine ylide, which can be generated *in situ* through several manners, reacts efficiently with maleimides, maleates and their derivatives, fumarates, acrylic systems, vinylic sulfones, nitroalkenes, enones, inones, etc.^[11] The main trend of all these acceptors is that a carbon-carbon double or triple bond are involved in the pericyclic transformation affording the corresponding [3+2] cycloaddition product. However, it has been reported that some examples with particular alkenes did not follow the general [3+2]

pattern. For example, a [3+3] cycloaddition was observed in self-cyclizations of azomethine ylides derived from isatin with various primary or cyclic secondary amines,^[12] with quinone monoimides,^[13] or using phthalazinium dicyanomethanides.^[14] [3+6] Cycloadditions were detected employing 2-acylcycloheptatrienes,^[15] fulvenes,^[16] or tropone,^[17] and also 1,7-electrocyclizations have been published.^[18, 19] In addition, highly functionalized 1,2,4,5-tetrazepine derivatives have been obtained via unprecedented [4+3] cycloaddition of in situ generated azoalkenes with C,N-cyclic azomethine imine.^[20] However, [4+3] cycloaddition processes involving azomethine ylides have not been reported yet.

Looking for an expansion of the scope of this 1,3-DC, 1,2-diaza-1,3-dienes (DDs) **1** (Scheme 1) were envisaged as dipolarophile candidates. The chemical properties of DDs are based on to the electron-withdrawing effect of the azo group in the heterodiene system. DDs **1** present an umpolung^[21] of the traditional carbonyl reactivity.^[22] This nature makes these neutral compounds enable to nucleophilic additions at the terminal carbon atom of the azo-ene system. Thus, the reactivity of DDs **1** remains opposite with respect to the natural polarity of the known carbonyl derivatives, and their employment opens the possibility to have a carbon atom in the α -position to the carbonyl group with a reverse of the normal polarity. Besides, the substituents of this system play an important role, and electron-withdrawing groups on the terminal carbon or nitrogen atom enhance the stability and the electrophilic character of the heterodiene system.^[23] DDs **1** are important synthetic intermediates and versatile building blocks in the construction of a large number of heterocycles.^[24] They are good Michael acceptors in 1,4-addition reactions^[25] and can react as dienes in inverse electron-demand [4+2] cycloadditions to give tetrahydropyridazines.^[26] DDs **1** have been also submitted to 1,3-DC in combination with nitrones to afford both 3-substituted-5-diazenyl isoxazolidines,^[27] and 2,3,4,7-tetrahydro-1,2,4,5-oxatriazepine.^[28]

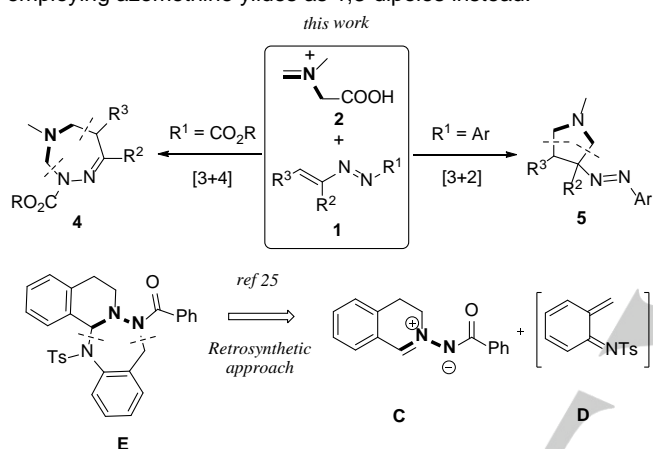
To the best of our knowledge, DDs have not been used as dipolarophiles with azomethine ylides generated by thermal decarboxylation of **2**, so in this work the reactivity of several azomethine ylides with these azo-ene derivatives will be surveyed. On the basis of the substituents on the 1,2-diaza-1,3-dienes a diversity-oriented synthesis (DOS)^[29] can be designed

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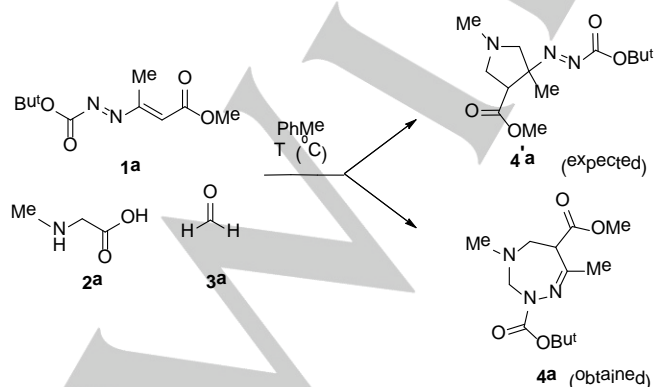
according to the different skeleton that would be achieved (Scheme 1). So, this divergent synthesis would afford through a [3+2] cycloaddition pyrrolidines **5**, which are demonstrated to be biologically active compounds and very interesting building blocks in synthetic organic chemistry.^[30] The same reaction would furnish also 1,2,4-triazepines **4** through a [3+4] cycloaddition that are not so common in the literature as their preparation is difficult.^[31] They belong to the family of triazepines^[32] which are very attractive, both from the pharmacological and chemical points of view. Recently, 1,2,4-triazepines **E** were prepared by 1,3-DC of *C*-*N*-cyclic azomethine imines **C** and aza-*o*-quinonodimethanes **D** which provided the 1,2-dinitrogen component.^[33] In our work, the 1,2-dinitrogenated fragment would be incorporated by DDs **1** employing azomethine ylides as 1,3-dipoles instead.



Scheme 1. Diversity-oriented synthesis of triazepines **4** or pyrrolidines **5** starting from DDs **1** and azomethine ylides **2**. Comparison of triazepine's assembly pathways by means of 1,3-dipolar cycloadditions.

Results and Discussion

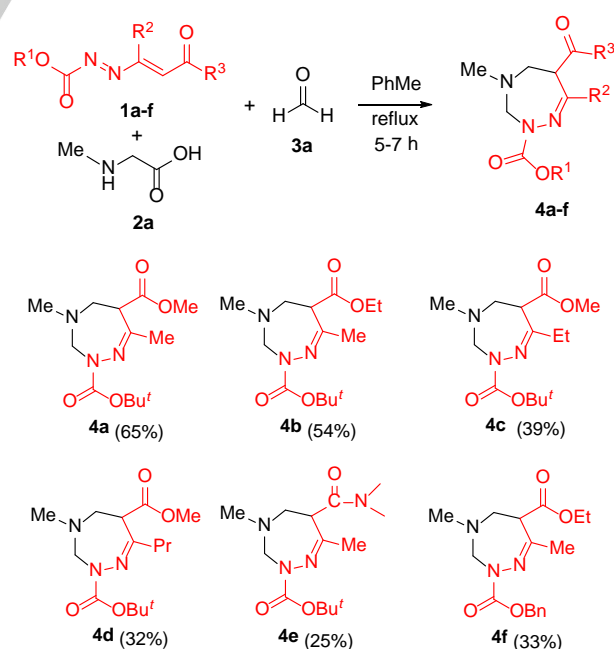
Based on our experience we decided to develop a conventional thermal multicomponent 1,3-DC using azomethine ylides and 1,2-diaza-1,3-dienes **1** as dipolarophiles.



Scheme 2. Multicomponent 1,3-DC between DD **1a**, sarcosine **2a**, and paraformaldehyde **3a**.

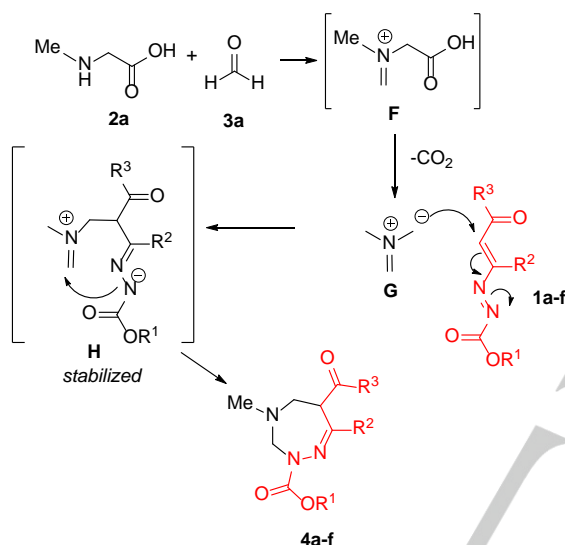
For the optimization of the process we employed as model DD **1a**,^[34] sarcosine **2a** as *N*-alkylated amino acid and paraformaldehyde **3a** (Scheme 2). While the reaction at room temperature and at 70 °C didn't work, by increasing the reaction temperature until the boiling point of toluene, the TLC analysis of the reaction revealed the formation of a new product that was isolated by flash chromatography. Initially, we expected the formation of pyrrolidine **4'a**, but, with our surprise, the NMR spectrum of the isolated product clearly indicates that 1,2,4-triazepine **4a** was achieved in 58% yield. In fact, the signals in the NMR carbon spectrum at 158.9, 170.9, and 174.8 ppm, respectively indicate the presence of an hydrazone moiety together with two ester functions, in agreement with the structure of 1,2,4-triazepine **4a**. The reaction was performed also in a microwave reactor at 100 °C for 2 h obtaining very low conversions (< 10%) and some decomposition products. An increment of the yield (65%) was achieved by adding two equiv of aldehyde **3a** and also two equivalents of the DD **1a**.

Under these optimal reaction conditions (Scheme 2), we next examined the reactivity of different *N*-alkylated amino acids and aldehydes. Unfortunately, the incorporation of other aldehydes such as cinnamaldehyde, ethyl glyoxylate, 2,2-dimethoxyacetaldehyde, crotonaldehyde, pentanal and benzaldehyde in the reaction of DD **1a** and sarcosine **2a** was unfruitful, even heating the reaction mixture at 140 °C in a pressure tube. Also when *N*-methylalanine was allowed to react with paraformaldehyde and DD **1a**, no reaction product was detected by ¹H NMR spectroscopy of the crude product. The scope of the reaction was next evaluated in the reaction of sarcosine **2a** with paraformaldehyde **3a** and different DDs **1a-f**. 1,2,4-Triazepines **4a-f** were obtained moderate to good yield (30–65%) (Scheme 3).



Scheme 3. Substrate scope of different DDs **1a-f** in the reaction with sarcosine **2a**, and paraformaldehyde **3a**. Reaction conditions: DD **1a** (1.0 mmol), sarcosine **2a** (0.5 mmol), paraformaldehyde **3a** (1.0 mmol), toluene (5.0 mL), reflux.

A plausible mechanism for this [4+3] cycloaddition is described in Scheme 4. Initially, the azomethine ylide **G** is generated by a thermal decarboxylation of the iminium intermediate **F**. Then, assuming that a more or less stepwise mechanism takes place, a Michael addition of the azomethine ylides **G** to the terminal carbon atom of the azo-ene system of DDs **1a-f** generates the hydrazone anion intermediates **H**. The subsequent *N*-intramolecular nucleophilic attack on the iminium moiety furnishes the final 1,2,4-triazepines **4a-f**.

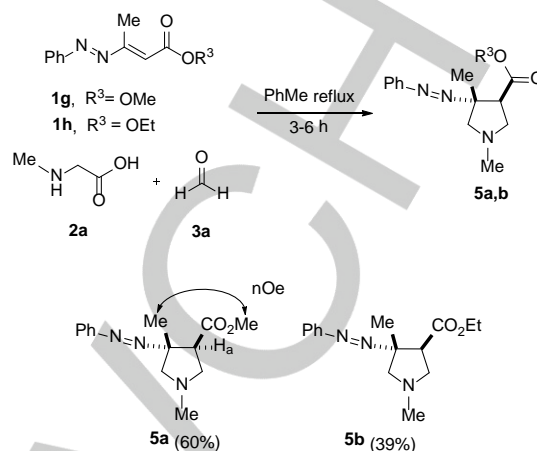


Scheme 4. Plausible mechanism for the multicomponent 1,3-DC between DDs **1a-f**, sarcosine **2a**, and paraformaldehyde **3a**.

The presence of an electron-withdrawing group on to the terminal nitrogen of the azo-ene system (CO_2R^1) in *N*-EWG-DDs **1a-f** seems to be crucial for the reaction course, as it stabilizes the intermediate **H**. Aware of this possibility, we have tested the reaction of sarcosine **2a**, paraformaldehyde **3a** and *N*-phenyl-DDs **1g,h** bearing a non-withdrawing substituent like a phenyl group at the nitrogen in position 1 of the azo-ene system, under the same conditions previously employed.

With our delight, we have noted that this simple change in the structure of the DDs **1** produces a different reaction pattern, affording a conventional [3+2] dipolar cycloaddition (Scheme 5). In these cases, *N*-phenyl-DDs **1g** and **1h** afforded the corresponding 4-(phenyldiazenyl)pyrrolidine-3-carboxylates **5a** and **5b** in 60 and 39% yields, respectively. The signals of NMR carbon spectra have been diagnostic to reveal the structure of compounds **5**. The signal of a quaternary carbon at around 80.0 ppm and the lack of the hydrazone moiety confirm the formation of the pyrrolidinic core. The relative configuration was unambiguously confirmed by nOe experiments, detecting a negative nOe between methyl group and H_a , but a large positive one between both methyl groups in molecule **5a**. This

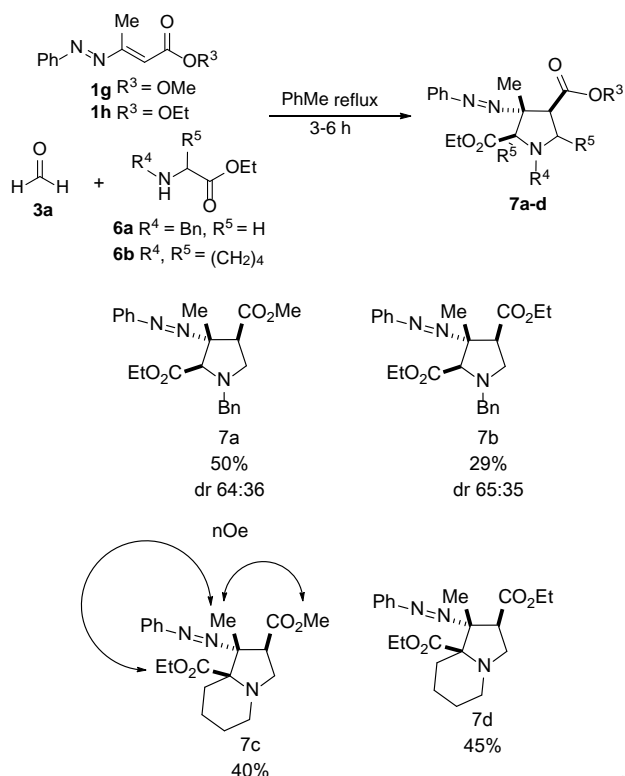
stereochemical outcome derives from the high stereospecificity of the 1,3-DC which justify a concerted process at this temperature.



Scheme 5. [3+2] Dipolar cycloaddition of *N*-phenyl-DDs **1g,h** in the reaction with sarcosine **2a**, and paraformaldehyde **3a**. Reaction conditions: DDs **1g,h** (1.0 mmol), sarcosine **2a** (0.5 mmol), paraformaldehyde **3a** (1.0 mmol), toluene (5.0 mL), reflux.

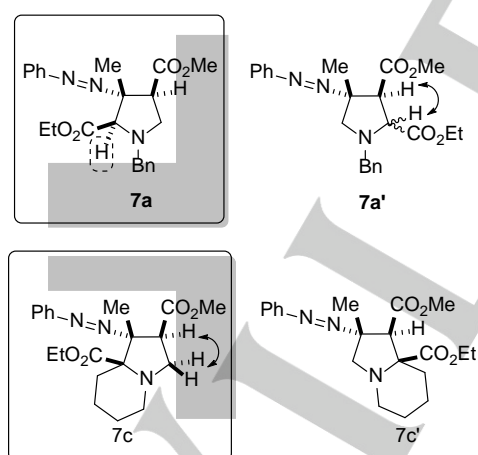
The different behavior observed in the 1,3-DC reaction of *N*-phenyl-DDs **1g,h**, encouraged us to incorporate other aldehydes such as ethyl glyoxylate, benzaldehyde. While all these attempts carried out with the aldehydes were unfruitful, the reaction between *N*-phenyl-DDs **1g,h**, *N*-alkylated amino esters such as *N*-benzyl glycine ethyl ester **6a** or ethyl pipercolinate **6b** and paraformaldehyde **3a** in toluene under reflux furnished the corresponding 1-benzyl 3-(phenyldiazenyl)pyrrolidine-2,4-dicarboxylates **7a,b** and 1-(phenyldiazenyl)octahydroindolizine-2,8a-dicarboxylates **7c,d**. (Scheme 6). In these latter cases, the generation of azomethine ylide occurs through a [1,2]-prototropy shift after the formation of the iminium salt with paraformaldehyde **3a**.

The structure of **7a-d** was confirmed by NMR methods: the absence of signals due to hydrazone moiety in ^{13}C NMR, and quaternary carbons at around 80 ppm indicates that [3+2] dipolar cycloaddition occurred. Again, nOe experiments of compound **7c** revealed the *cis*-arrangement of both esters and methyl groups. With respect to the regioselectivity of the attack of the negative charge of the dipole, starting from *N*-EWG-DDs **1a-f**, the addition occurred with the opposite trend observed for the reactions involving *N*-phenyl-DDs **1g,h** (Scheme 7). In the case of 1-benzyl 3-(phenyldiazenyl)pyrrolidine-2,4-dicarboxylate **7a**, the singlet signal of one hydrogen in ^1H -NMR at 4.58 ppm excludes the regioisomer **7a'**. For compound **7c**, the structure of the regioisomer obtained is confirmed by the ABX coupling pattern occurring between a methyne and a methylene group. The assignments were further confirmed by ^1H - ^1H COSY spectrum.



Scheme 6. [3+2] Dipolar cycloaddition of *N*-phenyl-DDs **1g,h**, paraformaldehyde **3a**, and glycine ethyl ester **6a** or ethyl pipercolinate **6b**.^[a]

^[a] Reaction conditions: *N*-phenyl-DDs **1g,h** (1.0 mmol), *N*-alkylated amino esters **6a,b** (0.5 mmol), paraformaldehyde **3a** (1.0 mmol), toluene (5.0 mL), reflux.



Scheme 7. Possible regioisomers of 1-benzyl (phenyldiazenyl)pyrrolidine-dicarboxylate **7a,7a'** and (phenyldiazenyl)octahydroindolizine-dicarboxylate **7c, 7c'**.

A very important detail of this divergent synthesis was the influence of the terminal substituent in the azo component. In these last reactions, non-electron-withdrawing group present in *N*-phenyl-DDs **1g,h** favored the conventional attack to the α,β -unsaturated ester moiety affording pyrrolidines or

octahydroindolizine **5** and **7**, respectively, unlike to the reactivity exhibited by EWG-DDs **1a-f** in which an electron-withdrawing substituent is bound at the terminal nitrogen of the azo-ene system. The chemical behaviour of these different DDs can be explained according to their different LUMO-coefficients.^[34] As depicted in Figure 1, the very large LUMO-coefficients of the EWG-DDs **1a** located in α -carbon to the ester group and the nitrogen atom of the aza-moiety interact with large coefficients of fleeting azomethine ylide **2a** (coefficients around 0.7 and -0.7 in the two carbon atoms). Note the most stable transoid conformation of DD **1a**, at high temperature it must be converted in its corresponding cisoid one to complete the reaction. Again we can propose a concerted mechanism due the conservation of the relative configuration of the substituents of the DD.

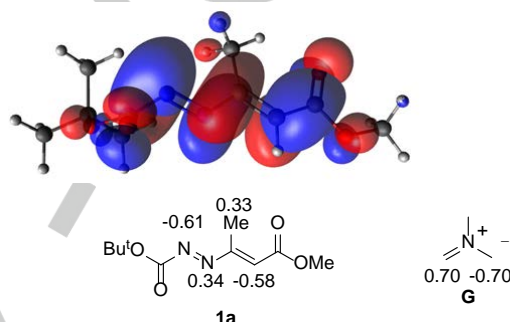


Figure 1. LUMO orbital coefficients for **1a** calculated at B3LYP/6-311+g(d,p) level of theory.

In contrast, the LUMO-coefficients for *N*-phenyl-DDs **1g-h** show very similar size. The value of α -carbon to the ester group is -0.5, and this is the main reactive centre; similarly, the contiguous carbon atom has a 0.40 value. As five membered rings are formed faster than seven membered ones, the [3+2] occurred preferentially rather than the [4+3] (Figure 2). We presumed that a step-wise process occur rather than a concerted one.

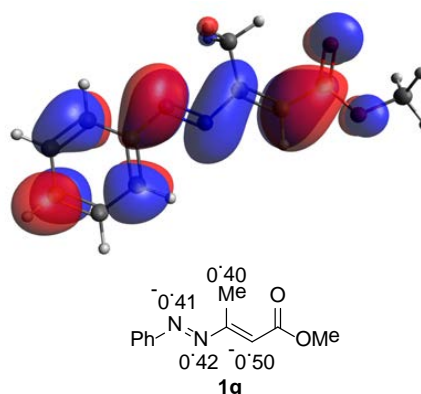


Figure 2. LUMO orbital coefficients for **1g** calculated at B3LYP/6-311+g(d,p) level of theory for the two conformers.

Conclusions

We have reported the first 1,3-DC between azomethine ylides and 1,2-diaza-1,3-dienes (DDs). Two main classes of products can be obtained: 1,2,4-triazepines, and pyrrolidines by means of [3+4] or [3+2] cycloadditions, respectively. We have demonstrated as the preference of the [3+4] cycloaddition is controlled by the presence in *N*-EWG-DDs **1a-f** of an electron-withdrawing group bound directly to the azo-moiety able to stabilize the anionic intermediate. On the contrary, under the same conditions the reactions of *N*-phenyl DDs lacking of electron withdrawing moiety on the nitrogen in position 1 of the azo-ene system gave the [3+2] cycloaddition product. Furthermore, the [3+2] cycloadditions show the same regioselectivity of the attack of the negative charge of the dipole displayed for the reactions involving nitrones,^[20] but with the opposite trend in respect to the [3+4] cycloaddition.

The normal *endo*-approach of DDs **1** was confirmed in the [3+2] processes finding two ester and methyl groups in a *cis*-arrangement. Considering the operational simplicity and that the obtained products represent interesting scaffolds deployed with a variety of functional groups, a convenient and practical synthetic methodology has been developed.

Experimental Section

General information. All the commercially available reagents and solvents were used without further purification. 1,2-Diaza-1,3-dienes **1a-h** were synthesized as a mixture of *E/Z* isomers as previously reported.³⁵ Sarcosine **2a**, *N*-benzyl glycine ethyl ester **6a** or ethyl pipercolinate **6b** and paraformaldehyde **3a** are commercial materials and were used without further purification. Chromatographic purification of compounds was carried out on silica gel (60–200 μ m). TLC analysis was performed on pre-loaded (0.25 mm) glass supported silica gel plates (Kieselgel 60); compounds were visualized by exposure to UV light and by dipping the plates in 1% Ce(SO₄) \cdot 4H₂O, 2.5% (NH₄)₆Mo₇O₂₄ \cdot 4H₂O in 10% sulphuric acid followed by heating on a hot plate. All ¹H NMR and ¹³C NMR spectra were recorded at 400 and 100.56 MHz, respectively. Proton and carbon spectra were referenced internally to solvent signals, using values of δ = 2.50 ppm for proton (middle peak) and δ = 39.50 ppm for carbon (middle peak) in DMSO-*d*₆ and δ = 7.27 ppm for proton and δ = 77.00 ppm for carbon (middle peak) in CDCl₃. The following abbreviations are used to describe peak patterns where appropriate: s = singlet, d = doublet, t = triplet, q = quartet, sex = sextet, m = multiplet and br = broad signal. All coupling constants (*J*) are given in Hz. FT-IR spectra were obtained as Nujol mulls. Mass spectra were recorded in the EI mode (70 eV). Melting points were determined in open capillary tubes and are uncorrected.

General procedure for the synthesis of 3,4,5,6-tetrahydro-2H-1,2,4-triazepine-2,6-dicarboxylate 4a-f. DDs **1a-f** (2.0 mmol), sarcosine **2a** (90 mg, 1.0 mmol), and paraformaldehyde **3a** (60 mg, 2.0 mmol) were dissolved in toluene (5 mL). The reactions were refluxed for 5.0–7.0 h. Then, the solvent was evaporated under reduced pressure and the crude mixture was chromatographed on silica gel (ethyl acetate : cyclohexane) obtaining pure 3,4,5,6-tetrahydro-2H-1,2,4-triazepine-2,6-dicarboxylates **4a-f**.

2-Tert-butyl 6-methyl 4,7-dimethyl-3,4,5,6-tetrahydro-2H-1,2,4-triazepine-2,6-dicarboxylate 4a: Pale yellow oil, 185 mg, 65%; IR ν_{\max} : 1758, 1713 cm⁻¹; ¹H-NMR δ : 1.50 (s, 9H, C(CH₃)₃), 2.07 (s, 3H, CH₃), 2.42 (s, 3H, NCH₃), 2.85 (dd, 1H, *J* = 13.2 Hz, *J* = 2.4 Hz, CHCH₂NCH₃),

3.07 (dd, 1H, *J* = 13.2 Hz, *J* = 7.6 Hz, CHCH₂NCH₃), 3.73–3.79 (m, 1H, CHCH₂NCH₃), 3.75 (s, 3H), 4.23 (d, 1H, *J* = 14.0 Hz, NCH₂N), 4.72 (d, 1H, *J* = 14.0 Hz, NCH₂N); ¹³C-NMR δ : 23.6 (q), 28.1 (q), 41.2 (q), 49.0 (t), 51.7 (d), 52.4 (q), 69.6 (t), 81.6 (s), 158.9 (s), 170.9 (s), 174.8 (s); MS (EI) *m/z* (%): 285 (M⁺, 1%), 270 (3), 254 (2), 226 (2), 212 (7), 169 (10), 153 (18), 110 (9); anal. calcd. for C₁₃H₂₃N₃O₄ (285.3395): C 54.72, H 8.12, N 14.73; found: C 54.59, H 8.08, N 14.87.

2-Tert-butyl 6-ethyl 4,7-dimethyl-3,4,5,6-tetrahydro-2H-1,2,4-triazepine-2,6-dicarboxylate 4b: Pale yellow oil, 161 mg, 54%; IR ν_{\max} : 1758, 1695 cm⁻¹; ¹H-NMR (CDCl₃) δ : 1.28 (t, 3H, *J* = 7.2 Hz, OCH₂CH₃), 1.50 (s, 9H, C(CH₃)₃), 2.08 (s, 3H, CH₃), 2.42 (s, 3H, NCH₃), 2.84 (dd, 1H, *J* = 13.2 Hz, *J* = 2.4 Hz, CHCH₂NCH₃), 3.05 (dd, 1H, *J* = 13.2 Hz, *J* = 7.6 Hz, CHCH₂NCH₃), 3.75 (dd, 1H, *J* = 7.6 Hz, *J* = 2.0 Hz, CHCH₂NCH₃), 4.17–4.25 (m, 3H, OCH₂CH₃, NCH₂N), 4.73 (d, 1H, *J* = 14.0 Hz, NCH₂N); ¹³C-NMR δ : 14.0 (q), 23.6 (q), 28.15 (q), 41.2 (q), 49.1 (t), 51.8 (d), 61.4 (t), 69.8 (t), 81.4 (s), 154.4 (s), 159.1 (s), 170.4 (s); MS (EI) *m/z* (%): 299 (M⁺, 1%), 254 (7), 241 (30), 225 (18), 198 (28), 153 (10), 125 (38); anal. calcd. for C₁₄H₂₅N₃O₄ (299.3661): C 56.17, H 8.42, N 14.04; found: C 56.29, H 8.45, N 13.92.

2-Tert-butyl 6-methyl 7-ethyl-4-methyl-3,4,5,6-tetrahydro-2H-1,2,4-triazepine-2,6-dicarboxylate 4c: Pale yellow oil, 117 mg, 39%; IR ν_{\max} : 1748, 1697 cm⁻¹; ¹H-NMR δ : 1.14 (t, 3H, *J* = 7.6 Hz, CH₂CH₃), 1.50 (s, 9H, C(CH₃)₃), 2.29–2.45 (m, 2H, CH₂CH₃), 2.39 (s, 3H, NCH₃), 2.80 (dd, 1H, *J* = 12.8 Hz, *J* = 2.4 Hz, CHCH₂NCH₃), 3.12 (dd, 1H, *J* = 12.8 Hz, *J* = 6.8 Hz, CHCH₂NCH₃), 3.72 (dd, 1H, *J* = 6.8 Hz, *J* = 2.4 Hz, CHCH₂NCH₃), 3.75 (s, 3H, OCH₃), 4.34 (d, 1H, *J* = 14.0 Hz, NCH₂N), 4.57 (d, 1H, *J* = 14.0 Hz); ¹³C-NMR δ : 11.0 (q), 28.2 (q), 30.8 (t), 41.6 (q), 48.6 (t), 52.3 (d), 52.5 (q), 69.6 (t), 81.3 (s), 154.4 (s), 162.4 (s), 171.0 (s); MS (EI) *m/z* (%): 299 (M⁺, 2%), 268 (2), 226 (6), 198 (4), 183 (10), 167 (17); anal. calcd. for C₁₄H₂₅N₃O₄ (299.3661): C 56.17, H 8.42, N 14.04; found: C 56.06, H 8.39, N 14.27.

2-Tert-butyl 6-methyl 4-methyl-7-propyl-3,4,5,6-tetrahydro-2H-1,2,4-triazepine-2,6-dicarboxylate 4d: Pale yellow oil, 100 mg, 32%; IR ν_{\max} : 1750, 1704 cm⁻¹; ¹H-NMR δ : 0.93 (t, 3H, *J* = 7.2 Hz, CH₂CH₂CH₃), 1.48 (s, 9H, C(CH₃)₃), 1.59 (sex, 2H, CH₂CH₂CH₃), 2.22–2.35 (m, 2H, CH₂CH₂CH₃), 2.40 (s, 3H, NCH₃), 2.77 (dd, 1H, *J* = 12.8 Hz, *J* = 2.4 Hz, CHCH₂NCH₃), 3.10 (dd, 1H, *J* = 12.8 Hz, *J* = 6.8 Hz, CHCH₂NCH₃), 3.70 (dd, 1H, *J* = 6.8 Hz, *J* = 2.4 Hz, CHCH₂NCH₃), 3.73 (s, 3H, OCH₃), 4.32 (d, 1H, *J* = 14.0 Hz, NCH₂N), 4.52 (d, 1H, *J* = 14.0 Hz, NCH₂N); ¹³C-NMR δ : 14.1 (q), 21.0 (t), 28.2 (q), 39.4 (t), 41.4 (q), 48.7 (t), 52.2 (q), 52.4 (d), 69.5 (t), 81.2 (s), 161.9 (s), 170.9 (s), 171.1 (s); MS (EI) *m/z* (%): 298 (2), 281 (3), 255 (4), 197 (21), 181 (18), 167 (31), 153 (24); anal. calcd. for C₁₅H₂₇N₃O₄ (313.3927): C 57.49, H 8.68, N 13.41; found: C 57.38, H 8.66, N 13.52.

Tert-butyl 6-[(dimethylamino)carbonyl]-4,7-dimethyl-3,4,5,6-tetrahydro-2H-1,2,4-triazepine-2-dicarboxylate 4e: Pale yellow oil, 75 mg, 25%; IR ν_{\max} : 1746, 1707 cm⁻¹; ¹H-NMR δ : 1.50 (s, 9H, C(CH₃)₃), 2.02 (s, 3H, CH₃), 2.45 (s, 3H, NCH₃), 2.79–2.95 (m, 2H, CHCH₂NCH₃), 2.97 (s, 3H, N(Me)₂), 2.98 (s, 3H, N(Me)₂), 3.86 (d, 1H, *J* = 13.6 Hz, NCH₂N), 4.21 (dd, 1H, *J* = 9.2 Hz, *J* = 2.0 Hz, CHCH₂NCH₃), 4.87 (d, 1H, *J* = 13.6 Hz, NCH₂N); ¹³C-NMR δ : 21.6 (q), 28.2 (q), 35.5 (q), 37.4 (q), 40.4 (q), 44.4 (t), 51.7 (d), 69.6 (t), 81.4 (s), 154.5 (s), 166.0 (s), 170.1 (s); MS (EI) *m/z* (%): 298 (M⁺, 2%), 241 (4), 225 (19), 197 (63), 153 (18), 125 (78); anal. calcd. for C₁₆H₂₆N₄O₃ (298.3814): C 56.35, H 8.78, N 18.78; found: C 56.49, H 8.83, N 18.66.

2-Benzyl 6-ethyl 4,7-dimethyl-3,4,5,6-tetrahydro-2H-1,2,4-triazepine-2,6-dicarboxylate 4f: Pale yellow oil, 117 mg, 39%; IR ν_{\max} : 1746, 1704 cm⁻¹; ¹H-NMR δ : 1.28 (t, 3H, *J* = 7.2 Hz, OCH₂CH₃), 2.11 (s, 3H, CH₃), 2.44 (s, 3H, NCH₃), 2.86 (dd, 1H, *J* = 13.2 Hz, *J* = 2.4 Hz, CHCH₂NCH₃), 3.10 (dd, 1H, *J* = 13.2 Hz, *J* = 7.6 Hz, CHCH₂NCH₃), 3.79 (dd, 1H, *J* = 7.6 Hz, *J* = 2.4 Hz, CHCH₂NCH₃), 4.18–4.23 (m, 2H, OCH₂CH₃), 4.31 (d, 1H, *J* = 14.0 Hz), 4.81 (d, 1H, *J* = 14.0 Hz), 5.26 (s, 2H, OCH₂Ph), 7.30–7.42 (m, 5H, Ar); ¹³C-NMR δ : 14.0 (q), 23.7 (q), 41.2 (q), 49.2 (t), 51.7 (d), 61.5 (t), 68.0 (t), 69.9 (t), 128.1 (d), 128.3 (d), 128.4 (d), 136.1 (s), 160.5 (s), 166.5 (s), 170.2 (s); MS (EI) *m/z* (%): 287 (7), 260 (2), 167 (28), 149 (100), 123 (50); anal. calcd. for C₁₇H₂₃N₃O₄ (333.3382): C 61.25, H 6.95, N 12.60; found: C 61.38, H 6.99, N 12.47.

General procedure for the synthesis of 4-[phenyldiazanyl]pyrrolidine-3-carboxylates 5a,b 3-(2-

phenyldiazenyl)pyrrolidine-2,4-dicarboxylates 7a,b, 1-(phenyldiazenyl)hexahydroindolizine-2,8a(1*H*)-dicarboxylate 7c,d. Sarcosine **2a** (90 mg, 1.0 mmol), or *N*-benzylglycine ethyl ester **6a** (193 mg, 1.0 mmol), or ethyl pipercolinate **6b** (157 mg, 1.0 mmol), DDs **1g,h** (2.0 mmol), and paraformaldehyde **3a** (2.0 mmol) were dissolved in toluene (5 mL). The reactions were refluxed for 5.0–7.0. Then, the solvent was evaporated under reduced pressure and the crude mixture was chromatographed on silica gel (ethyl acetate : cyclohexane) obtaining 4-[phenyldiazenyl]pyrrolidine-3-carboxylates **5a,b**, 3-(2-phenyldiazenyl)pyrrolidine-2,4-dicarboxylates **7a,b**, 1-(phenyldiazenyl)hexahydroindolizine-2,8a(1*H*)-dicarboxylate **7c,d**.

Methyl 1,4-dimethyl-4-[phenyldiazenyl]pyrrolidine-3-carboxylate 5a: Pale yellow oil, 157 mg, 60%; IR ν_{max} : 1749 cm^{-1} ; $^1\text{H-NMR}$ δ : 1.32 (s, 3H, CH_3), 2.38 (s, 3H, NCH_3), 2.52 (d, 1H, $J = 10.0$ Hz, CH_2NCH_3), 2.92 (dd, 1H, $J = 9.6$ Hz, $J = 8.4$ Hz, $\text{CHCH}_2\text{NCH}_3$), 3.13 (dd, 1H, $J = 8.8$ Hz, $J = 8.4$ Hz, $\text{CHCH}_2\text{NCH}_3$), 3.41 (d, 1H, $J = 10.0$ Hz, CH_2NCH_3), 3.71 (s, 3H, OCH_3), 3.98 (t, 1H, $J = 8.4$ Hz, $\text{CHCH}_2\text{NCH}_3$), 7.40–7.48 (m, 3H, Ar), 7.74 (dd, 2H, $J = 7.6$ Hz, $J = 1.2$ Hz, Ar); $^{13}\text{C-NMR}$ δ : 19.8 (q), 42.1 (q), 49.5 (d), 51.6 (q), 58.1 (t), 66.6 (t), 79.7 (s), 122.5 (d), 128.8 (d), 130.6 (d), 151.4 (s), 173.4 (s); MS (EI) m/z (%): 261 (M^+ 2%), 183 (17), 169 (43), 156 (45), 149 (100); anal. calcd. for $\text{C}_{14}\text{H}_{19}\text{N}_3\text{O}_2$ (261.3197): C 64.35, H 7.33, N 16.08; found: C 64.46, H 7.36, N 15.97.

Ethyl 1,4-dimethyl-4-[phenyldiazenyl]pyrrolidine-3-carboxylate 5b: Pale yellow oil, 107 mg, 39%; IR ν_{max} : 1734 cm^{-1} ; $^1\text{H-NMR}$ δ : 1.26 (t, 3H, $J = 7.2$ Hz, OCH_2CH_3), 1.34 (s, 3H, CH_3), 2.38 (s, 3H, NCH_3), 2.51 (d, 1H, $J = 10.0$ Hz, CH_2NCH_3), 2.91 (dd, 1H, $J = 9.2$ Hz, $J = 9.2$ Hz, $\text{CHCH}_2\text{NCH}_3$), 3.13 (dd, 1H, $J = 8.8$ Hz, $J = 8.8$ Hz, $\text{CHCH}_2\text{NCH}_3$), 3.41 (d, 1H, $J = 10.0$ Hz, CH_2NCH_3), 3.98 (t, 1H, $J = 8.4$ Hz, $\text{CHCH}_2\text{NCH}_3$), 4.13–4.20 (m, 2H, OCH_2CH_3), 7.40–7.48 (m, 3H, Ar), 7.73 (dd, 2H, $J = 8.4$ Hz, $J = 2.0$ Hz, Ar); $^{13}\text{C-NMR}$ δ : 14.3 (q), 19.8 (q), 42.2 (q), 49.7 (d), 58.2 (t), 60.5 (t), 66.9 (t), 79.7 (s), 122.5 (d), 128.8 (d), 130.5 (d), 151.5 (s), 172.9 (s); MS (EI) m/z (%): 275 (M^+ 5%), 260 (4), 231 (10), 183 (90), 170 (100); anal. calcd. for $\text{C}_{15}\text{H}_{21}\text{N}_3\text{O}_2$ (275.3463): C 65.43, H 7.69, N 15.26; found: C 65.31, H 7.65, N 15.36.

2-Ethyl 4-methyl 1-benzyl-3-methyl-3-(2-phenyldiazenyl)pyrrolidine-2,4-dicarboxylate 7a:³⁶ mixture of diastereoisomers Pale yellow oil, 205 mg, 50%; IR ν_{max} : 1755 cm^{-1} ; $^1\text{H-NMR}$ (dr 59:41, * denotes minor diastereoisomer signals) δ : 1.15* (t, 3H, $J = 7.2$ Hz, OCH_2CH_3), 1.19 (t, 3H, $J = 7.2$ Hz, OCH_2CH_3), 1.39 (s, 3H, CH_3), 1.49* (s, 3H, CH_3), 2.97 (t, 1H, $J = 9.6$ Hz, CHCO_2Me), 3.02* (t, 1H, $J = 9.6$ Hz, CHCO_2Me), 3.57* (s, 2H, NCH_2Ar), 3.63 (s, 3H, OCH_3), 3.65* (s, 3H, OCH_3), 3.63–3.73 (m, 2H, NCH_2Ar), 3.89–3.96 (m, 1H, CH_2N), 4.03–4.22 (m, 4H, OCH_2CH_3 , CH_2N , CHCO_2Et), 7.26–7.39 (m, 5H, Ar), 7.44–7.52 (m, 3H, Ar), 7.68–7.74 (m, 2H, Ar).

2-Ethyl 4-methyl 1-benzyl-3-methyl-3-(2-phenyldiazenyl)pyrrolidine-2,4-dicarboxylate 7a: one pure diastereoisomer IR ν_{max} : 1755 cm^{-1} ; $^1\text{H-NMR}$ δ : 1.19 (t, 3H, $J = 7.2$ Hz, OCH_2CH_3), 1.38 (s, 3H, CH_3), 2.96 (t, 1H, $J = 9.6$ Hz, CHCO_2Me), 3.63 (s, 3H, OCH_3), 3.63–3.73 (m, 2H, NCH_2Ar), 3.92 (dd, 1H, $J = 9.2$ Hz, $J = 7.4$ Hz, CH_2N), 4.03–4.23 (m, 4H, OCH_2CH_3 , CH_2N , CHCO_2Et), 7.26–7.33 (m, 5H, Ar), 7.43–7.52 (m, 3H, Ar), 7.72 (dd, 2H, $J = 8.0$ Hz, $J = 1.6$ Hz, Ar); $^{13}\text{C-NMR}$ δ : 14.2 (q), 16.0 (q), 49.1 (t), 51.7 (d), 52.4 (q), 57.2 (t), 60.5 (t), 72.0 (d), 80.6 (s), 122.3 (d), 127.1 (s), 128.2 (d), 128.8 (d), 129.0 (d), 130.6 (d), 151.6 (s), 170.7 (s), 171.6 (s); MS (EI) m/z (%): 303 (15), 244 (20), 231 (7), 217 (5), 123 (34); anal. calcd. for $\text{C}_{23}\text{H}_{27}\text{N}_3\text{O}_4$ (409.4873): C 67.46, H 6.65, N 10.26; found: C 67.56, H 6.67, N 10.14.

Diethyl 1-benzyl-3-methyl-3-(2-phenyldiazenyl)pyrrolidine-2,4-dicarboxylate 7b:³⁷ mixture of diastereoisomers Pale yellow oil, 123 mg, 29%; IR ν_{max} : 1742, 1752 cm^{-1} ; $^1\text{H-NMR}$ (dr 65:35, * denotes minor diastereoisomer signals) δ : 1.15* (t, 3H, $J = 7.2$ Hz, OCH_2CH_3), 1.16 (t, 3H, $J = 6.8$ Hz, OCH_2CH_3), 1.18 (t, 3H, $J = 7.2$ Hz, OCH_2CH_3), 1.21* (t, 3H, $J = 7.2$ Hz, OCH_2CH_3), 1.40 (s, 3H, CH_3), 1.50* (s, 3H, CH_3), 2.96 (t, 1H, $J = 10.0$ Hz, CHCO_2Me), 3.03* (t, 1H, $J = 9.6$ Hz, CHCO_2Me), 3.63–3.72 (m, 2H, NCH_2Ar), 3.86–3.91 (m, 1H, CH_2N), 4.02–4.22 (m, 6H, 2 OCH_2CH_3 , CH_2N , CHCO_2Et), 7.24–7.41 (m, 5H, Ar), 7.44–7.52 (m, 3H, Ar), 7.67–7.73 (m, 2H, Ar); $^{13}\text{C-NMR}$ δ : 14.1 (q), 14.2 (q), 15.7 (q), 19.8 (q), 48.2 (q), 49.5 (t), 52.3 (d), 53.1 (d), 57.4 (t), 57.6 (t), 60.4 (t), 60.5 (t), 60.6 (t), 72.1 (s), 75.0 (d), 80.4 (s), 122.2 (d), 122.3 (d), 122.5 (d), 127.1 (s), 127.2 (s), 128.1 (d), 128.2 (d), 128.8 (d), 128.8 (d), 128.9 (d), 129.2 (d), 130.5 (d), 130.7 (d), 151.4 (s), 151.7 (s), 170.3 (s), 170.9 (s), 172.1

(s) 172.8 (s); MS (EI) m/z (%): 318 (4), 245 (5), 244 (16), 167 (17), 123 (46); anal. calcd. for $\text{C}_{24}\text{H}_{29}\text{N}_3\text{O}_4$ (423.5049): C 68.06, H 6.90, N 9.92; found: C 67.95, H 6.87, N 10.03.

8a-Ethyl 2-methyl 1-methyl-1-(phenyldiazenyl)hexahydroindolizine-2,8a(1*H*)-dicarboxylate 7c: Pale yellow oil, 149 mg, 40%; IR ν_{max} : 1754, 1752 cm^{-1} ; $^1\text{H-NMR}$ δ : 1.23 (t, 3H, $J = 7.2$ Hz, OCH_2CH_3), 1.30 (s, 3H, CH_3), 1.58–1.81 (m, 4H, CH_2CH_2), 1.97–2.06 (m, 2H, CH_2CH_2), 2.91 (dd, 1H, $J = 13.6$ Hz, $J = 4.4$ Hz, NCH_2CH_2), 3.13 (dt, 1H, $J = 13.2$ Hz, $J = 3.6$ Hz, NCH_2CH_2), 3.36 (dd, 1H, $J = 10.0$ Hz, $J = 9.6$ Hz, CHCO_2Me), 3.59 (s, 3H, CO_2CH_3), 3.81 (dd, 1H, $J = 9.2$ Hz, $J = 7.2$ Hz, $\text{CH}_2\text{CHCO}_2\text{Me}$), 4.13–4.27 (m, 3H, OCH_2CH_3 , $\text{CH}_2\text{CHCO}_2\text{Me}$), 7.43–7.49 (m, 3H, Ar), 7.70 (dd, 2H, $J = 8.0$ Hz, $J = 1.6$ Hz, Ar); $^{13}\text{C-NMR}$ δ : 14.2 (q), 16.9 (q), 20.2 (t), 21.8 (t), 27.1 (t), 45.6 (t), 46.6 (d), 50.2 (t), 51.5 (q), 60.6 (t), 75.9 (s), 81.1 (s), 122.4 (d), 128.9 (d), 130.4 (d), 151.8 (s), 172.1 (s), 172.3 (s); MS (EI) m/z (%): 300 (4), 195 (25), 164 (14), 136 (40); anal. calcd. for $\text{C}_{20}\text{H}_{27}\text{N}_3\text{O}_4$ (373.4462): C 64.32, H 7.29, N 11.25; found: C 64.45, H 7.34, N 11.17.

Diethyl 1-methyl-1-(phenyldiazenyl)hexahydroindolizine-2,8a(1*H*)-dicarboxylate 7d: Pale yellow oil, 174 mg, 45%; IR ν_{max} : 1743, 1736 cm^{-1} ; $^1\text{H-NMR}$ δ : 1.13 (t, 3H, $J = 7.2$ Hz, OCH_2CH_3), 1.20 (t, 3H, $J = 7.2$ Hz, OCH_2CH_3), 1.31 (s, 3H, CH_3), 1.64–1.89 (m, 4H, CH_2CH_2), 2.08 (d, 1H, $J = 13.2$ Hz, CH_2CH_2), 2.92 (d, 1H, $J = 10.8$ Hz, CH_2CH_2), 3.17 (dt, 1H, $J = 13.2$ Hz, $J = 3.2$ Hz, NCH_2CH_2), 3.37 (t, 1H, $J = 10.0$ Hz, NCH_2CH_2), 3.81–3.85 (m, 1H, CHCO_2Et), 4.00–4.26 (m, 6H, 2 OCH_2CH_3 , $\text{CH}_2\text{CHCO}_2\text{Me}$), 7.42–7.50 (m, 3H, Ar), 7.69 (dd, 2H, $J = 8.0$ Hz, $J = 1.6$ Hz, Ar); $^{13}\text{C-NMR}$ δ : 14.1 (q), 14.2 (q), 16.6 (q), 20.0 (t), 21.9 (t), 27.2 (t), 45.7 (d), 46.8 (t), 50.0 (t), 60.5 (t), 60.7 (t), 76.1 (s), 81.0 (s), 122.4 (d), 128.9 (d), 130.4 (d), 152.0 (s), 171.7 (s), 172.2 (s); MS (EI) m/z (%): 315 (3), 314 (13), 287 (7), 209 (42), 179 (12), 164 (14), 136 (82); anal. calcd. for $\text{C}_{21}\text{H}_{29}\text{N}_3\text{O}_4$ (387.4728): C 65.09, H 7.54, N 10.84; found: C 65.22, H 7.47, N 10.73.

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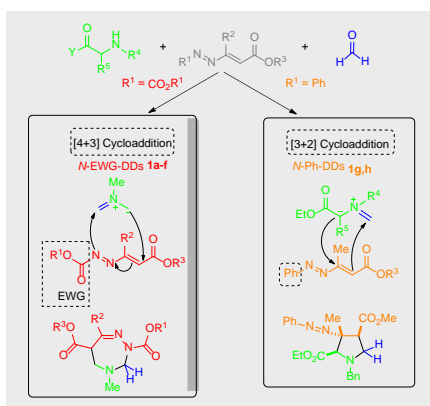
Keywords: Azomethine ylides • Cycloaddition • 1,2-Diaza-1,3-diene • 1,2,4-Triazepine • Pyrrolidine

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**[4+3] vs [3+2] cycloadditions**

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Reactivity of 1,2-diaza-1,3-dienes with
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